

Synthesis of α,β -Unsaturated Carbonyl Compounds by Successive Rearrangement and Peterson Olefination of β' -Hydroxy- α,β -epoxysilanes

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The reaction of β' -hydroxy- α,β -epoxysilanes with boron trifluoride-diethyl ether results in rearrangement of the epoxysilane skeleton followed by Peterson olefination, leading to α,β -unsaturated carbonyl compounds.

Rearrangement reactions of α,β -epoxysilanes and 1,2-elimination reactions of β -functionalized organosilanes have been extensively studied and widely applied to organic synthesis.¹ However, there are few investigations of the chemistry of β' -functionalized α,β -epoxysilanes which contain both an α,β -epoxysilane and a β -functionalized organosilane skeleton adjoining each other. The only reactions so far

reported are the 1,2-elimination reaction of β' -chloro- α,β -epoxysilanes by fluoride ion which leads to allene oxides,² and the fluoride-induced protodesilylation of β' -hydroxy- α,β -epoxysilanes.³

Recently we reported a general and convenient method for the preparation of β -hydroxyvinylsilanes (**1**) which was based on hydromagnesiation of silylacetylenes followed by treatment with aldehydes or ketones.⁴ Since the compounds (**1**) are readily epoxidized to β' -hydroxy- α,β -epoxysilanes (**2**) with *m*-chloroperbenzoic acid or VO(acac)₂-*t*-butyl hydroperoxide (acac = pentane-2,4-dione),⁵ we were interested in the behaviour of (**2**) under the conditions which would cause both rearrangement and 1,2-elimination.

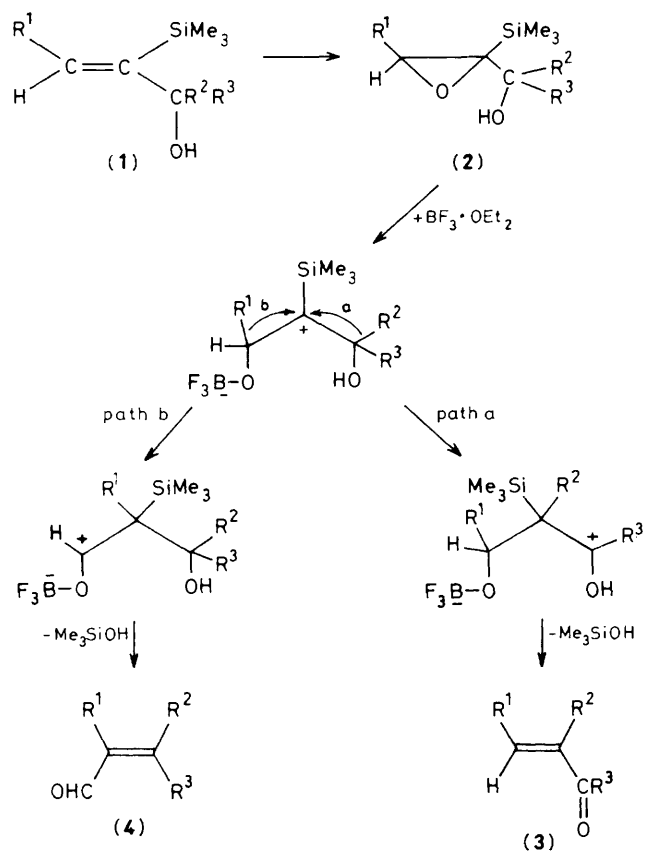
We have now found that when compounds (**2**) are treated with boron trifluoride-diethyl ether in methylene chloride, α,β -unsaturated carbonyl compounds (**3**) [and (**4**)] are obtained in good to excellent yields (Table 1). The production of (**3**) [and (**4**)], though rather strange at first glance, can be explained by assuming that rearrangement of the α,β -epoxysilane skeleton occurs prior to (and does not compete with) the 1,2-elimination reaction, as shown in Scheme 1. The step-wise pathway for the rearrangement process, which involves preliminary ring opening is shown in Scheme 1, but the concerted process could be an alternative.⁶

The structure of the products (**3**) [and (**4**)] indicated that the regio-chemistry of the rearrangement step was strongly

Table 1. Reaction of complexes (**2**) with BF₃·OEt₂.^a

Complex (2) ^b R ¹ = Bu ^a	Products ^{c,d}	Yield ^e (%)
a ; R ² = R ³ = H	(3a) ^f	58
b ; R ² = Me, R ³ = H	(3b), (4b) (1:1) ^g	83 ^h
c ; R ² = Bu, R ³ = H	(3c) ^f	82
d ; R ² = Ph, R ³ = H	(3d) ^f	85
e ; R ² = R ³ = Me	(3e), (4e) (3:1) ^g	90 ^h
f ; R ² -R ³ = -[CH ₂] ₅ -	(3f) ^f	75

^a BF₃·OEt₂ (3 equiv.) was added to a methylene chloride solution of (**2**) at room temperature and stirred for 15 min at that temperature. ^b Prepared by hydromagnesiation of 1-trimethylsilylhex-1-yne followed by treatment with the corresponding aldehyde or ketone. ^c Identified by comparison of ¹H n.m.r. spectra and g.l.c. retention times with those of authentic samples. ^d All products have (*E*)-configuration (by ¹H n.m.r.). ^e Isolated by distillation under reduced pressure. ^f >98% pure (by g.l.c. and ¹H n.m.r.). ^g Determined by g.l.c. ^h Combined yield.



Scheme 1

dependent on the substituents R^1 , R^2 , and R^3 in (2) individually, as well as in combination. With some compounds (2), a migration from $\text{C}(\beta')$ to $\text{C}(\alpha)$ occurred predominantly, thus making the reaction useful for the preparation of α,β -unsaturated carbonyl compounds such as (3a), (3c), (3d), and (3f). In other cases, the rearrangement involved the migration of R groups from both $\text{C}(\beta)$ and $\text{C}(\beta')$ to $\text{C}(\alpha)$, to form the mixtures of products (3b) and (4b), (3e) and (4e). However, in every case, no migration of H from $\text{C}(\beta)$ to $\text{C}(\alpha)$ was observed. It is also noteworthy that the products (3) [and (4)] thus obtained have (*E*)-configuration predominantly.

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